

# Heterogeneous catalytic decomposition of nitrous oxide.

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Nitrous oxide ( $\text{N}_2\text{O}$ ) is recognized as very harmful in destroying the stratospheric ozone layer. The Kyoto protocol of the United Nations Convention on Climate Change (December 1997) states that  $\text{N}_2\text{O}$  is a second non- $\text{CO}_2$  greenhouse gas. It has been reported that  $\text{N}_2\text{O}$  has 310 and 21 times the global warming potential of  $\text{CO}_2$  and  $\text{CH}_4$ , respectively.  $\text{N}_2\text{O}$  is also known to contribute to catalytic stratospheric ozone layer destruction.

The human contribution of the  $\text{N}_2\text{O}$  emission to the atmosphere is estimated to be 4.7-7 million ton per year, about 30-40% of the total emission. Some of the sources are nitric acid manufacturing, fossil fuels and biomass combustion, plus land cultivation. Unfortunately  $\text{N}_2\text{O}$  is also formed by selective catalytic reduction of  $\text{NO}_x$ . If the atmospheric  $\text{N}_2\text{O}$  is to be stabilised, there has to be a 70-80% reduction of the human emission. As it is now, there are no legislations on the emission of  $\text{N}_2\text{O}$ . It is though believed that it soon will come because of the growing governmental awareness, of the environmental impact of emission gasses. Emission reduction of  $\text{N}_2\text{O}$  can be achieved in different ways. One of them is after-treatment, end of pipe solutions. A possible after-treatment is catalytic decomposition of  $\text{N}_2\text{O}$  into the harmless gasses  $\text{N}_2$  and  $\text{O}_2$ . This reaction has been known for some time, and many catalysts have been tested for the decomposition. Still no promising catalyst has yet been found. For the catalyst to be suitable for industrial application, it needs to operate at relatively low temperature (under  $300^\circ\text{C}$ ), must not lose activity over time, have low cost in the making and of course have a high conversion of  $\text{N}_2\text{O}$ .

In this project, different catalysts have been prepared and analysed for their activity in this decomposition of  $\text{N}_2\text{O}$ , under conditions given by the industry. Among them are cesium doped cobalt-oxide spinels plus iron and copper modified zeolites. All the catalysts was analysed with different quantities of the impregnated material. This was done to find the optimum ratio of the support and for further analysing. The goal of this project is to find catalysts suitable for industrial application.